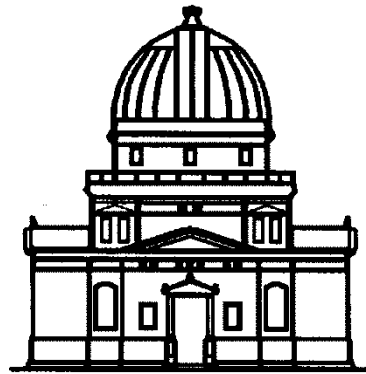


Introduction to Radioastronomy: Physical Processes



Observatoire astronomique
de Strasbourg

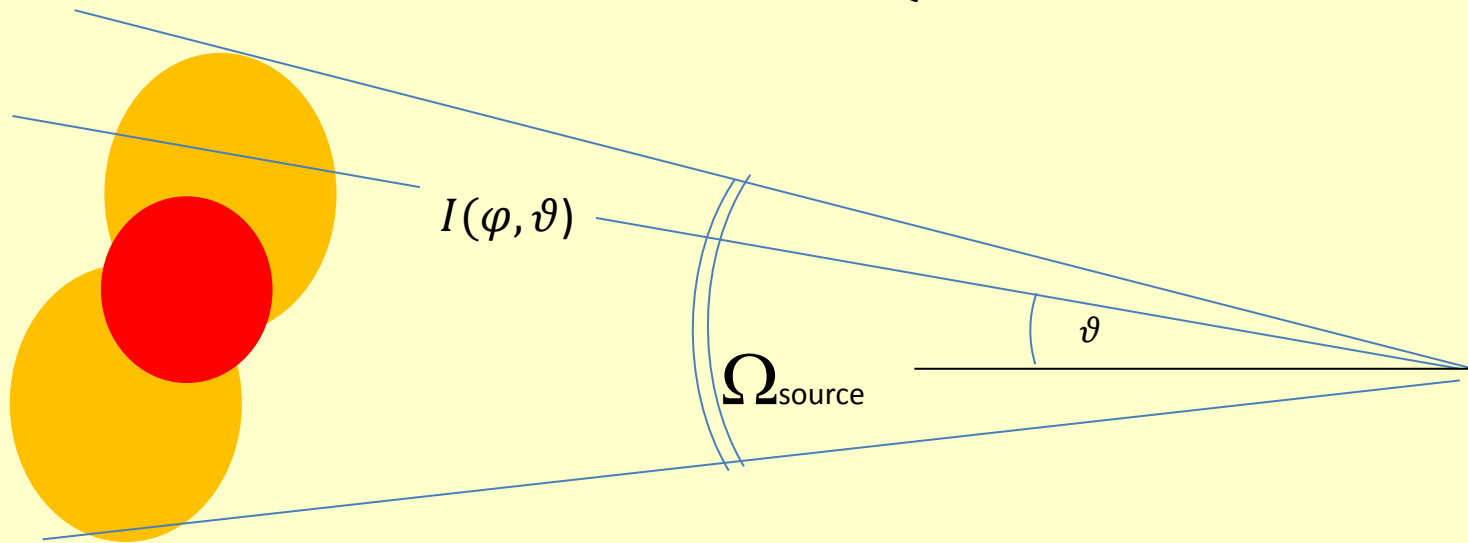
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<http://astro.u-strasbg.fr/~koppen/JKHome.html>

Radiation Quantities

- (specific) **Intensity** = energy flow per unit time, per unit area, per unit frequency, **into a certain direction, and per unit solid angle** [also used: Brightness, Surface brightness]; independent of distance if no extinction or refraction
- **Flux** = energy flow per unit time, per unit area, per unit frequency, and **perpendicular through the test area** [also: Flux density]; decreases with distance² from source

Radiation Quantities



Flux received from an extended source:

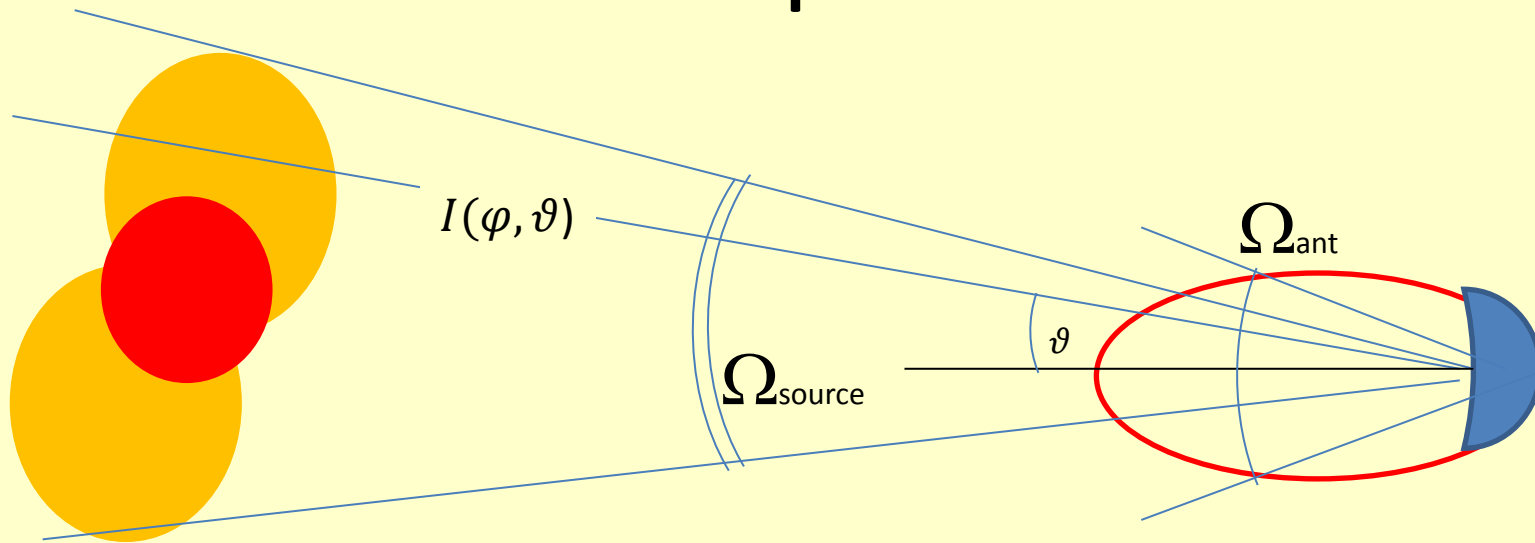
$$F = \int I(\varphi, \vartheta) \cos \vartheta \, d\varphi d\vartheta$$

homogeneous source: $F = I * \Omega_{\text{source}}$

'antenna pattern'
of test area

solid angle of spherical source: $\Omega_{\text{source}} = \pi r^2 / d^2$

The antenna pattern matters



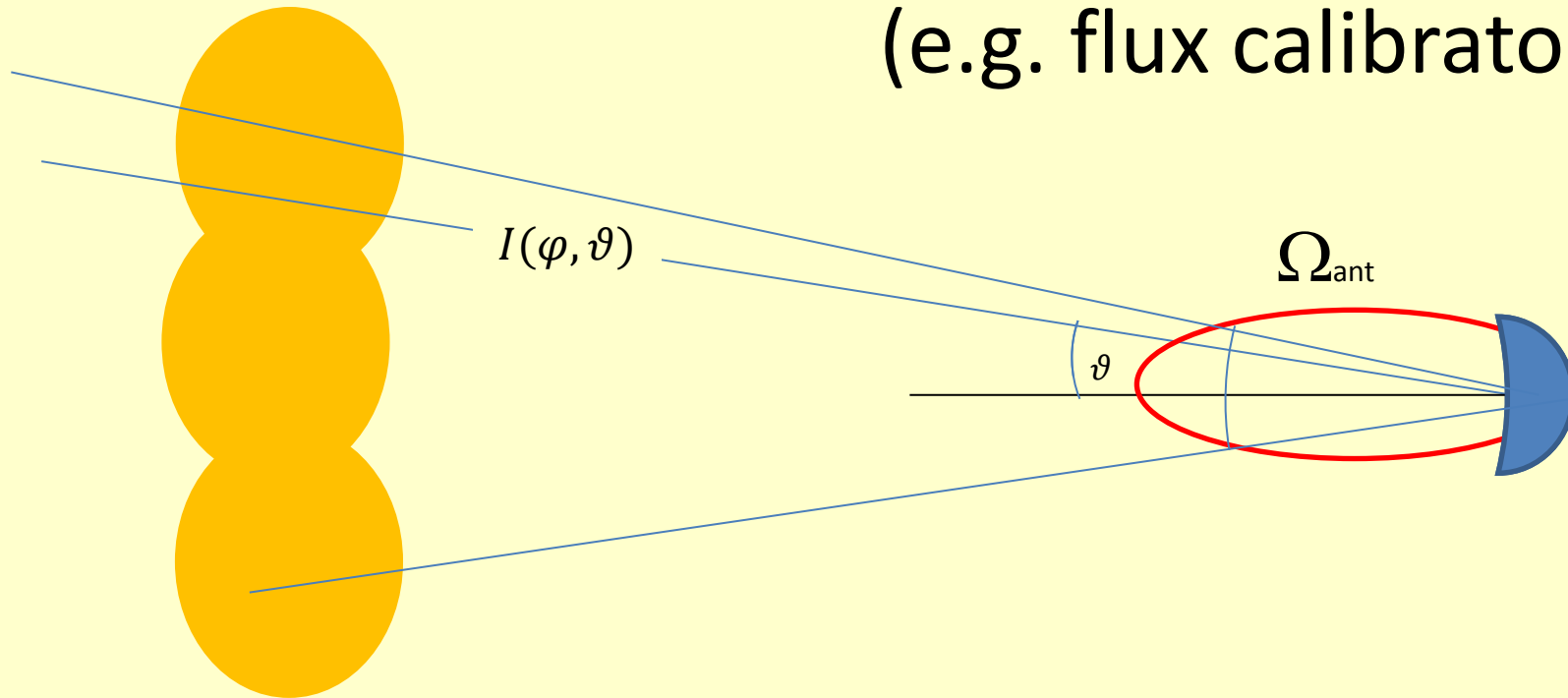
Flux received from source:

$$F = \int I(\varphi, \vartheta) A(\varphi, \vartheta) d\varphi d\vartheta$$

source brightness
distribution: Ω_{source}

antenna pattern
solid angle Ω_{ant}

Isotropic source ($\Omega_{source} \gg \Omega_{ant}$)
(e.g. flux calibrator)



$$\frac{2P}{A_{eff}} = F = I(T_B) * \int A(\varphi, \vartheta) d\varphi d\vartheta = I(T_B) * \Omega_{ant}$$

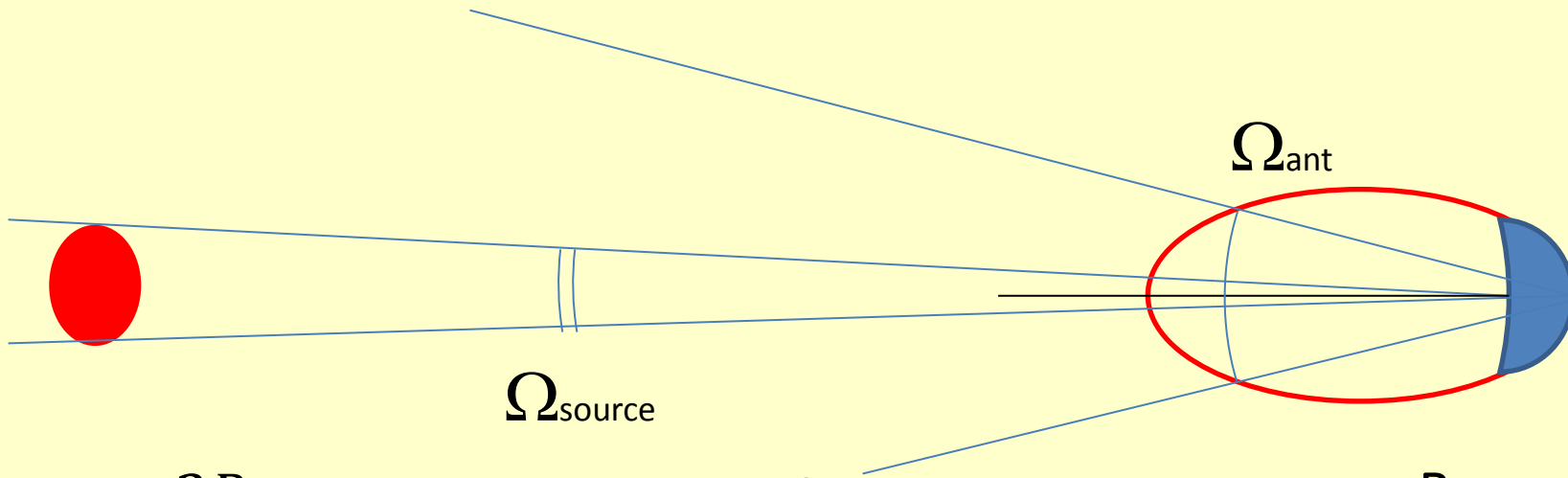
$$\frac{2k T_A}{\lambda^2 / \Omega_{ant}} = \frac{2k T_B}{\lambda^2} * \Omega_{ant}$$

$$T_A = T_B$$

antenna temperature

brightness temperature of source

Point source ($\Omega_{source} \ll \Omega_{ant}$)



$$\frac{2P}{A_{eff}} = F = A(\vartheta \approx 0) * \int I(\varphi, \vartheta) d\varphi d\vartheta$$

Beam centre:
 $A(\vartheta \approx 0) = 1$

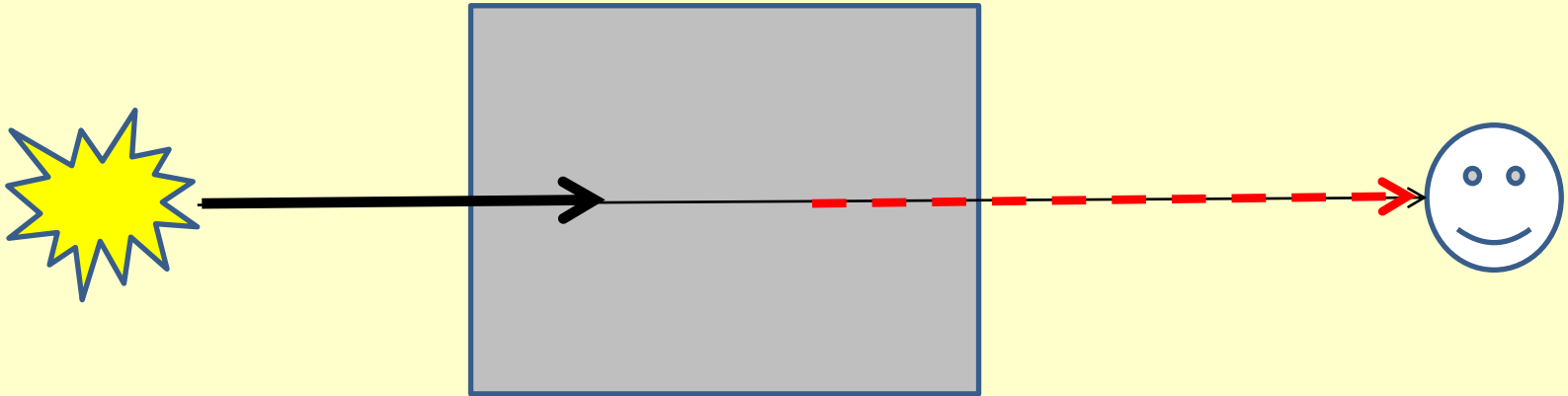
$$\frac{2k T_A}{\lambda^2 / \Omega_{ant}} = I(T_B) * \Omega_{source} = \frac{2k T_B}{\lambda^2} \Omega_{source}$$

$$T_A = T_B \frac{\Omega_{source}}{\Omega_{ant}} \leq T_B$$

Radiation and Matter

- Emissivity = energy production per unit time, unit volume, unit frequency
- Absorption coefficient (Opacity) = fraction of the incident radiative power absorbed, per unit volume, unit frequency

Radiative Transfer (I)



- Intensity is modified by absorption and emission:

$$\frac{dI}{ds} = -\kappa I + j$$

Indices of F , I , κ , j mark whether they are taken to be per unit frequency or per unit wavelength interval ...

Radiative Transfer (II)

- Formal solution (planar slab)
 - **Optical depth** (observer at $s=0$)

$$d\tau = -\kappa ds$$

$$\text{solution: } \tau(s) = \int_0^s \kappa(s') ds'$$

- Intensity

$$\begin{aligned} I &= I_0 e^{-\tau(L)} + \int_0^L j(s') e^{-\tau(s')} ds' \\ &= I_0 e^{-\tau(L)} + \int_0^L S(s') e^{-\tau(s')} d\tau(s') \end{aligned}$$

$$\text{Source function } S = j/\kappa \quad = B(f, T(s)) \quad (\text{in TE})$$

Radiative Transfer (IV)

- Homogeneous slab, **optically thick** limit:

$$\tau(L) \gg 1 :$$

$$I = 0 + S * \int_0^L e^{-\tau} d\tau = S(1 - e^{-\tau(L)}) \rightarrow S$$

$$T_B \rightarrow T_{\text{source}}$$

- ditto, **optically thin** limit:

$$\tau(L) \ll 1 \rightarrow \exp(-x) \approx 1-x :$$

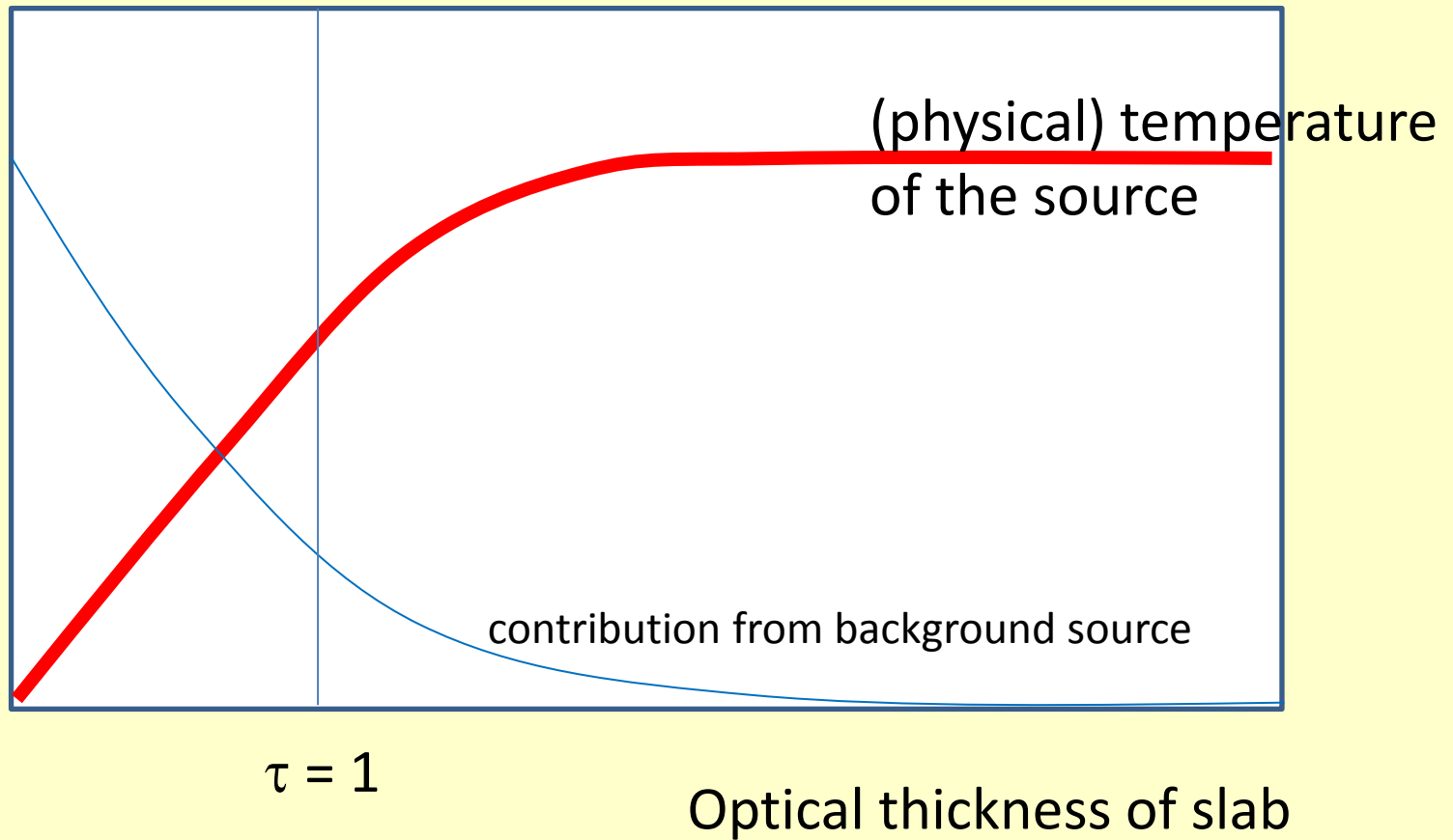
$$I \approx I_0 e^{-\tau(L)} + j * L = I_0 + S * \tau(L)$$

$$T_B \approx T_0 + T_{\text{source}} * \tau(L)$$

measures the amount of emitting matter

Radiative Transfer (IV)

Intensity
i.e.
brightness
temperature



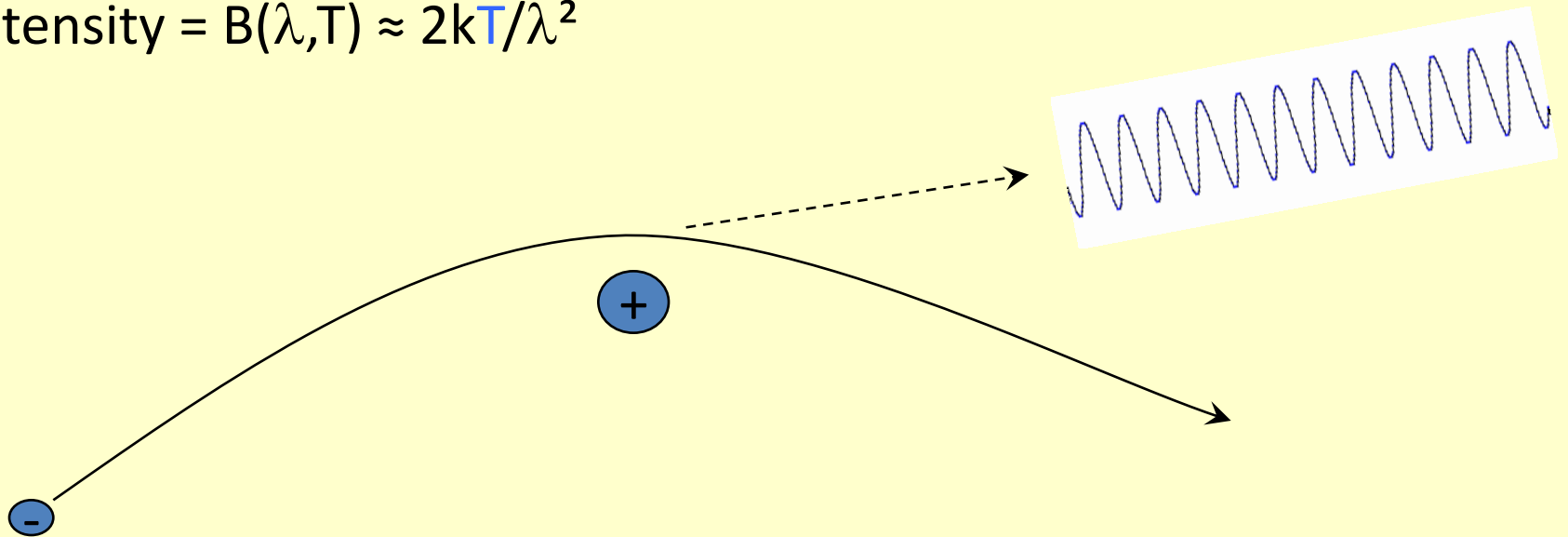
Physical processes that produce radio emission

- In principle: interaction of electromagnetic wave with charged particles
- **Continuum emission:** acceleration or deceleration (braking) of charged particles (electrons, protons)
- **Line emission:** downward transition of an electron between two bound states in an atom or molecule

Thermal emission (Bremsstrahlung):

by the random thermal motions of electrons passing by other charged particle (no polarization)

from opaque bodies: blackbody spectrum, at radio frequencies:
intensity = $B(\lambda, T) \approx 2kT/\lambda^2$



Radiation by accelerated charges

Single charge q with acceleration a : power emitted into unit solid angle:

$$\frac{dP}{d\Omega} = \frac{q^2}{4\pi c^3} a^2 \frac{\sin^2 \theta}{(1 - \beta \cos \theta)^5}$$

moving with speed v ($\beta = v/c$) parallel to acceleration, angle θ between line of sight and velocity vector

(Liénard 1898, $v \ll c$: Larmor 1897)

Gas of thermal electrons

- Maxwell distribution of velocities ($\varepsilon = mv^2/2$)

$$n(\varepsilon) = \frac{2N}{\sqrt{\pi} (kT)^{3/2}} \sqrt{\varepsilon} e^{-\frac{\varepsilon}{kT}}$$

- Emissivity (cgs units: $\text{erg s}^{-1} \text{cm}^{-3} \text{Hz}^{-1} \text{sterad}^{-1}$)

$$j_f(f) = N_e N_i \frac{8}{3} \sqrt{\frac{2\pi m}{3 kT}} \frac{Z^2 e^6}{m^2 c^3} g(f, T) e^{-\frac{hf}{kT}}$$

Gaunt factor $g(f, T)$

details: K.R.Lang, Astrophysical Formulae

The inverse process: absorption

- Kirchhoff's law (1860)

$$j_f = \kappa_f * B_f(f, T)$$

- → Absorption coefficient (cgs units: cm^{-1})

$$\kappa_f \approx \frac{0.009786 N_e N_i}{f^2 T^{\frac{3}{2}}} \ln\left(4.954 \cdot 10^7 \left(\frac{T^{3/2}}{Zf}\right)\right) \quad T < 316000 \text{ K}$$

$$\kappa_f \approx \frac{0.009786 N_e N_i}{f^2 T^{\frac{3}{2}}} \ln(4.954 \cdot 10^{10} T/f) \quad T > 316000 \text{ K}$$

Note: absorption increases with wavelength

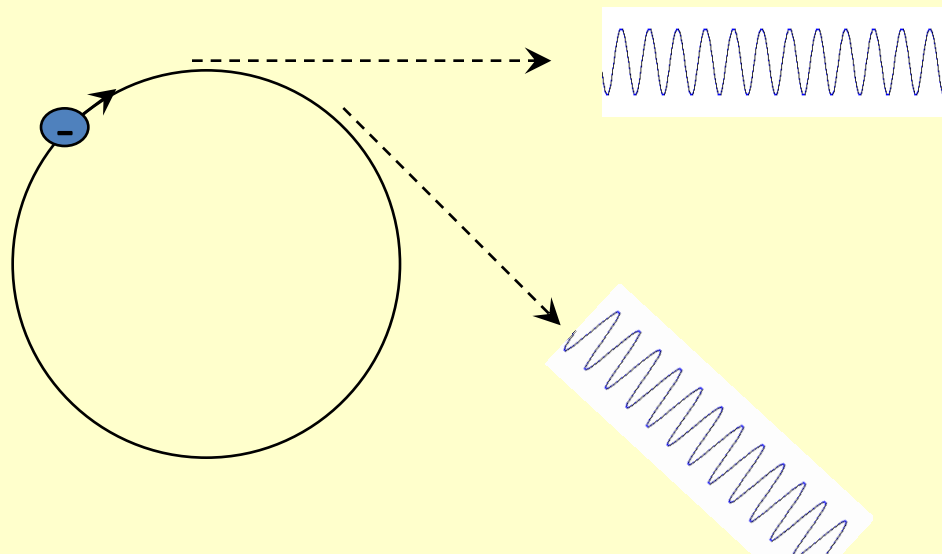
Synchrotron emission

Deflection of electrons in a magnetic field

→ electrons spiral around the magnetic field lines and emit a pulse whenever their motion is directed towards us

→ superposition of contributions by all electrons in the plasma ((thermal) distribution of speeds) gives:

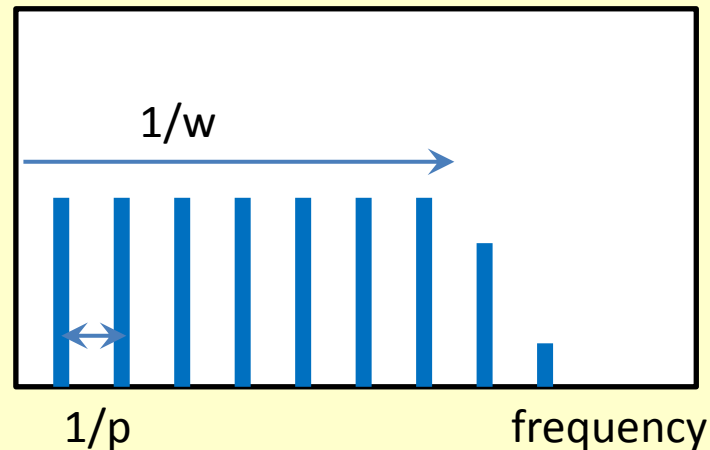
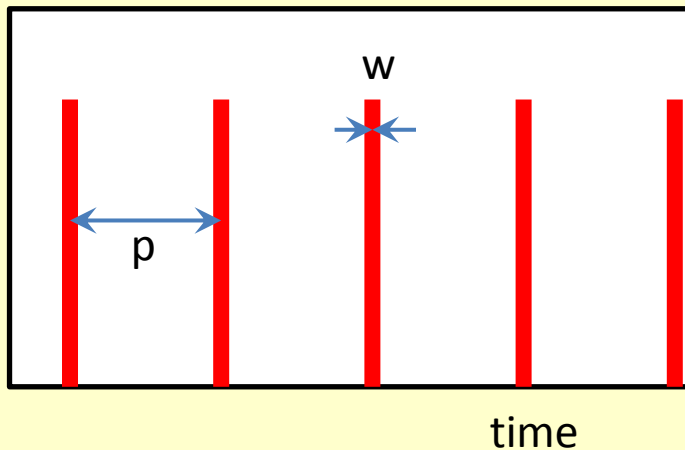
continuous spectrum (decreases with frequency, high frequency cutoff, linear polarization)



Synchrotron emission

At high speeds the distribution becomes a narrow cone with half width $\theta = \gamma^{-1} = \sqrt{1 - \beta^2}$ (from Liénard's equation, see above)

Electron in circular orbit of radius ρ with period $p = v/\rho \rightarrow$ pulse width $w = \gamma^{-3}/p \rightarrow$ observer sees harmonics of the rotation frequency $1/p$, up to the maximum number γ^3



Synchrotron emission

If one considers a population of electrons, distributed in energies like a power law: $dN/dE = \text{cte} * E^x$

it can be shown ... ☺ ...

that the frequency spectrum of the emitted radiation follows a power law with exponent $-(x-1)/2$

details: K.R.Lang, Astrophysical Formulae

Continuum spectra

log(Flux)



Synchrotron emission
(opt.thin)

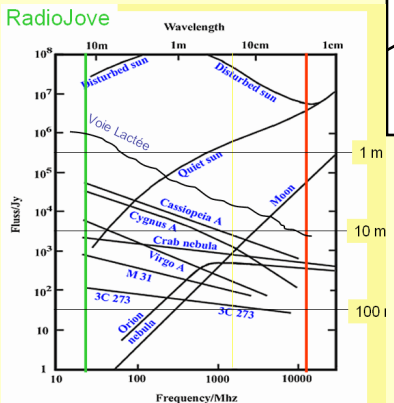
Optically thick: blackbody,
stars, Sun, Earth, humans

Thermal emission

Optically thick

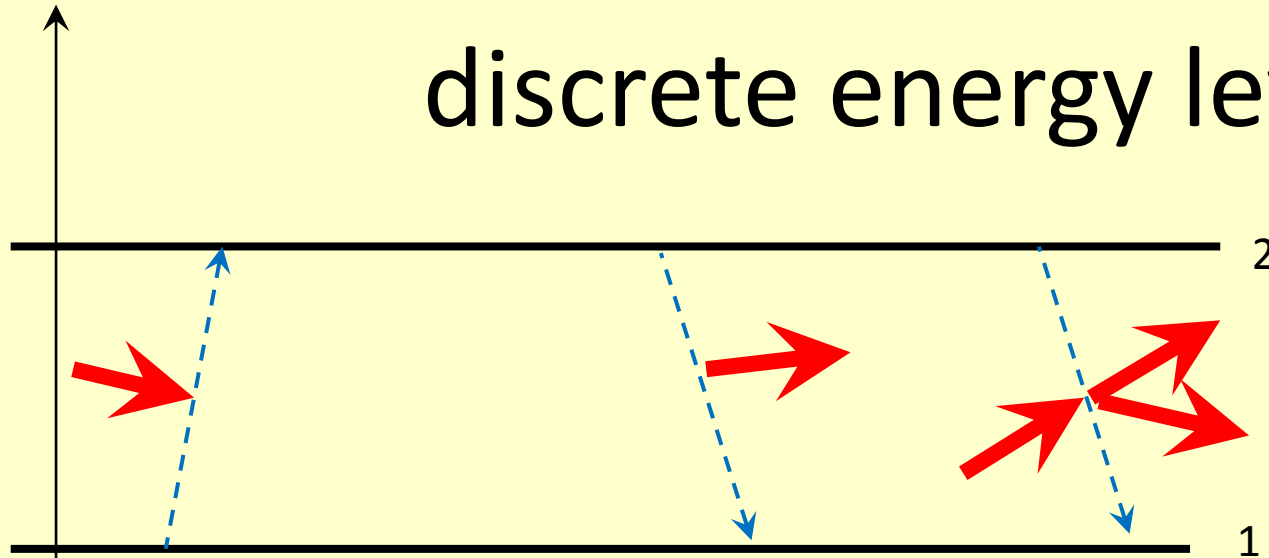
Optically thin: diffuse nebula

log(Frequency)



Lines = transitions between discrete energy levels

energy



Absorption

$$\text{rate} = n_1 B_{12} * J_{12}$$

Spontaneous
Emission

$$\text{rate} = n_2 A_{21}$$

Stimulated
Emission

$$\text{rate} = n_2 B_{21} * J_{12}$$

Einstein coefficients:

$$g_1 B_{12} = g_2 B_{21} \quad g_i = \text{statistical weight of level } i$$

$$A_{21} = B_{21} * 2hf^3/c^2 = \text{transition probability}$$

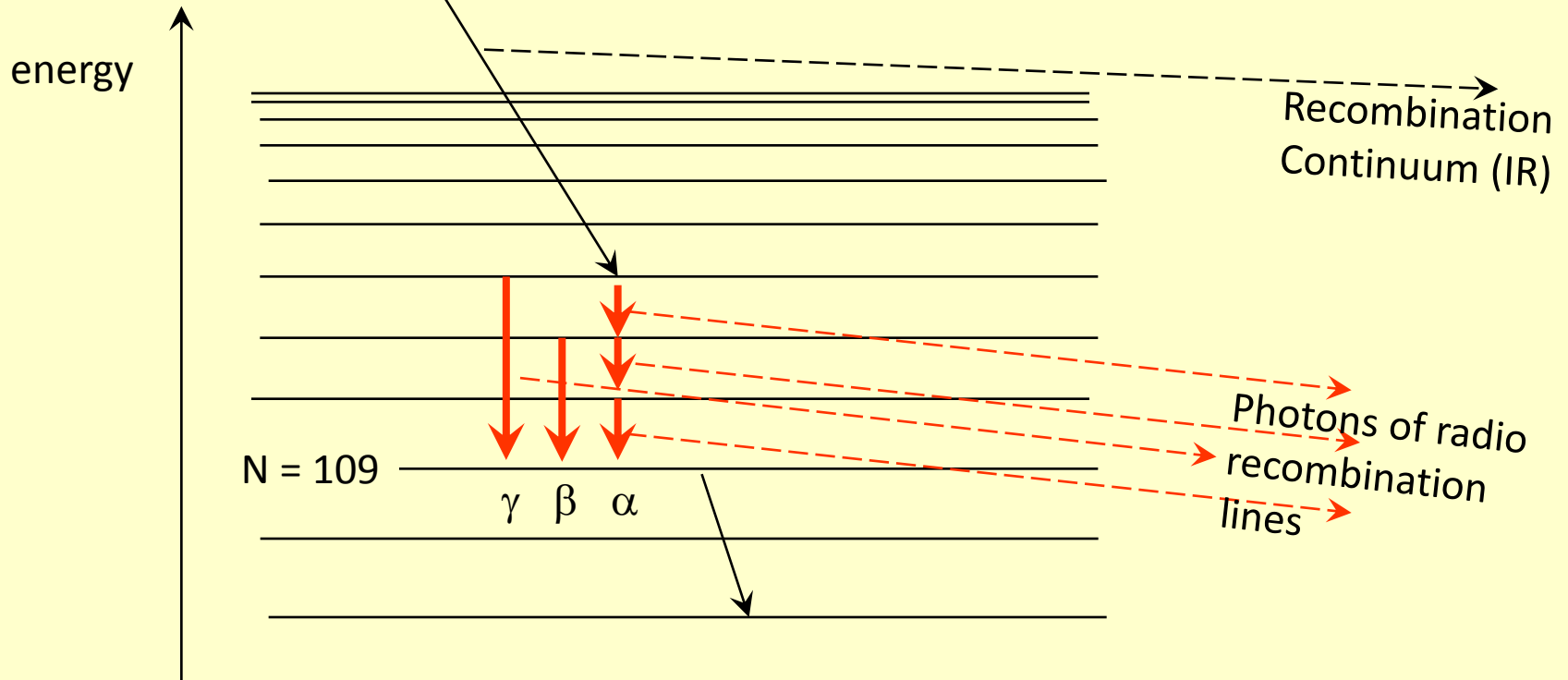
= inverse lifetime of upper state

$$\kappa_{12} = hf_{12}/4\pi * (n_1 B_{12} - n_2 B_{21}) * \varphi(f) \quad \text{absorp.coeff. with line profile function } \varphi$$

$$j_{21} = hf_{12}/4\pi * n_2 A_{21} * \varphi(f) \quad \text{emissivity}$$

Line emission (I)

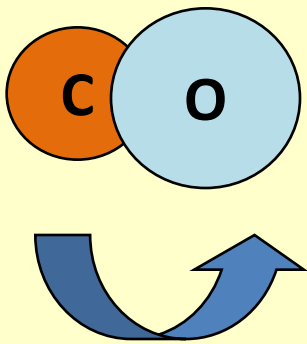
- Electron in hydrogen atom flips its spin (21 cm line at 1420.406 ... MHz, $A = 2.85 \cdot 10^{-15} \text{ s}^{-1}$ (magn.dipole))
- Electron changes energy state in atom (mostly hydrogen after recombination of proton and electron in plasma: recombination lines, H109 α 5 GHz, $A \approx 10^8 \text{ s}^{-1}$ (electr.dipole))



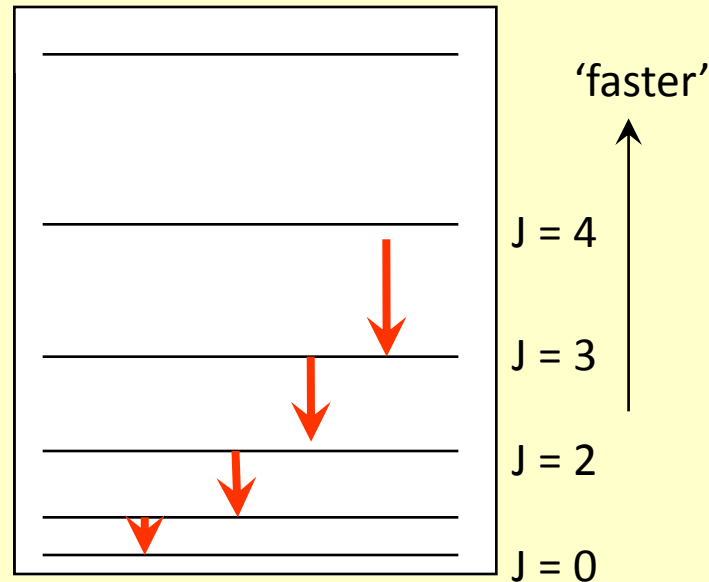
Line emission (II)

Molecule slows down from its high rotational state (into which it had been excited e.g. by a collision with another molecule (H_2))

CO = polar molecule



energy



Examples: CO lines

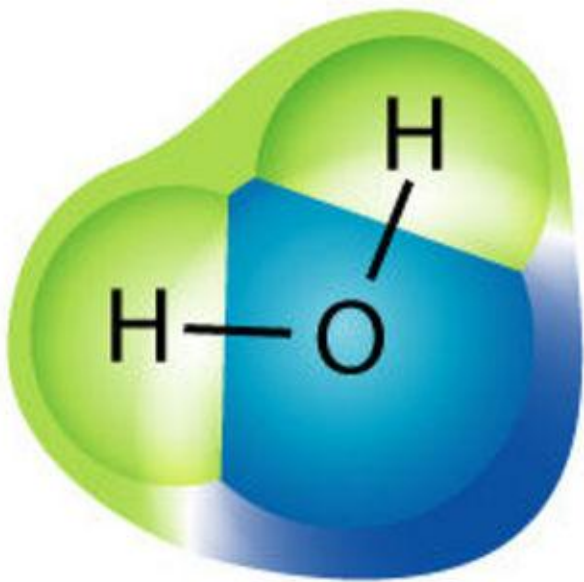
at 110, 220, 330, ... GHz

$A \approx 10^{-7} \text{ s}^{-1}$

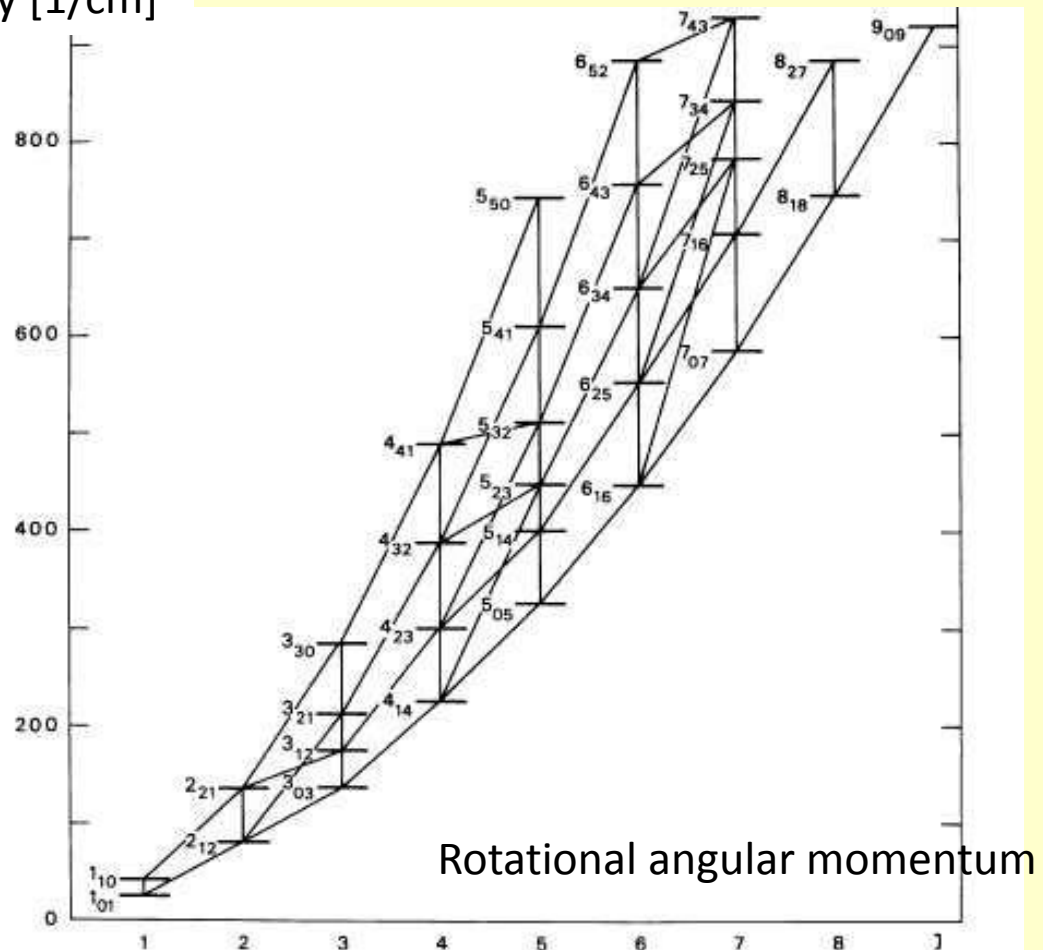
Line emission (III)

Water (asymmetric polar molecule)

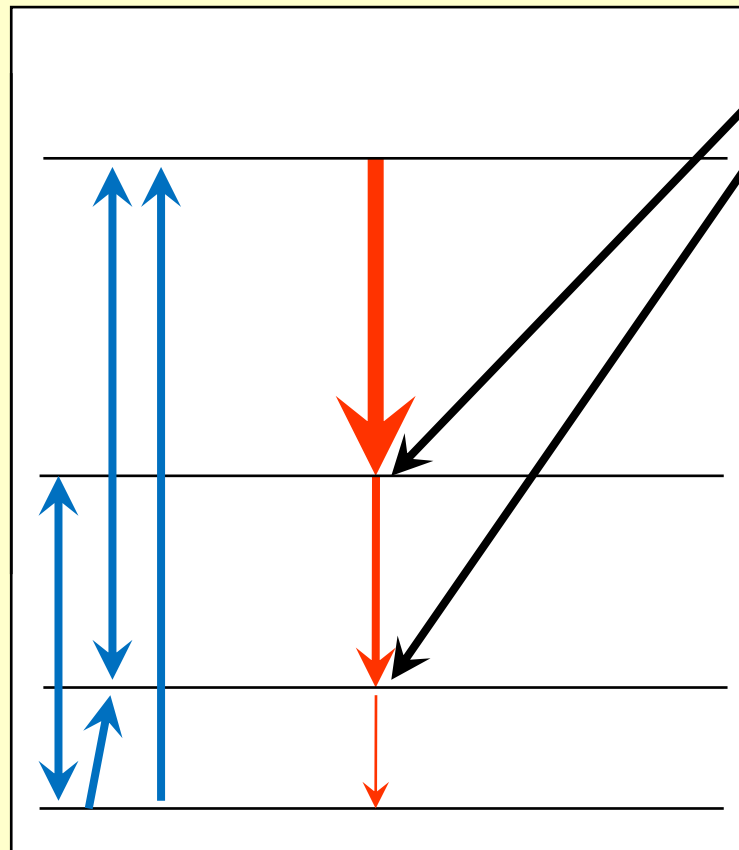
22 GHz



energy [1/cm]



Line emission (IV): Masers



energy

collisions

'traffic jam' leads to overpopulation of upper levels $n_2 > n_1$

→ **negative** absorption coefficient

$$\kappa_{12} = hf_{12}/4\pi * (n_1B_{12} - n_2B_{21}) * \varphi(f)$$

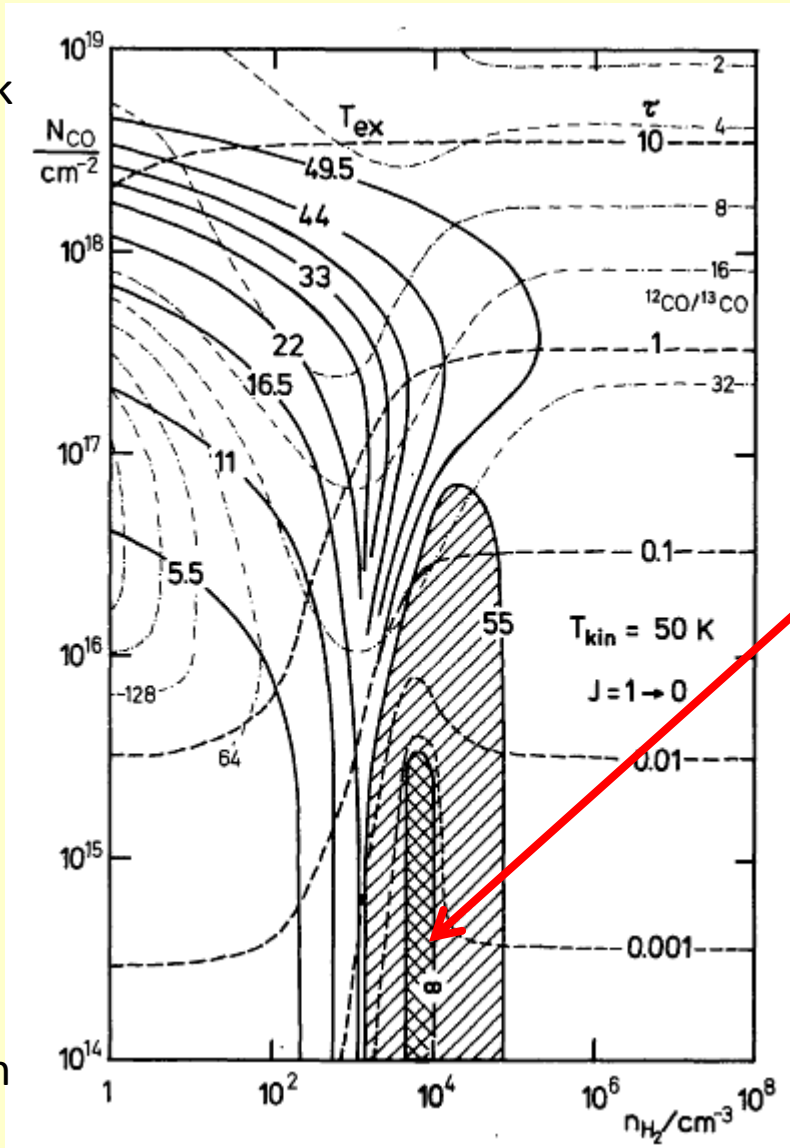
→ Incoming radiation is amplified

Examples:

- OH 1.7 GHz
- H₂O 22 GHz
- SiO 43 GHz
- CH₃OH 6.6 and 12 GHz

Line emission (V): Masers

Optically thick



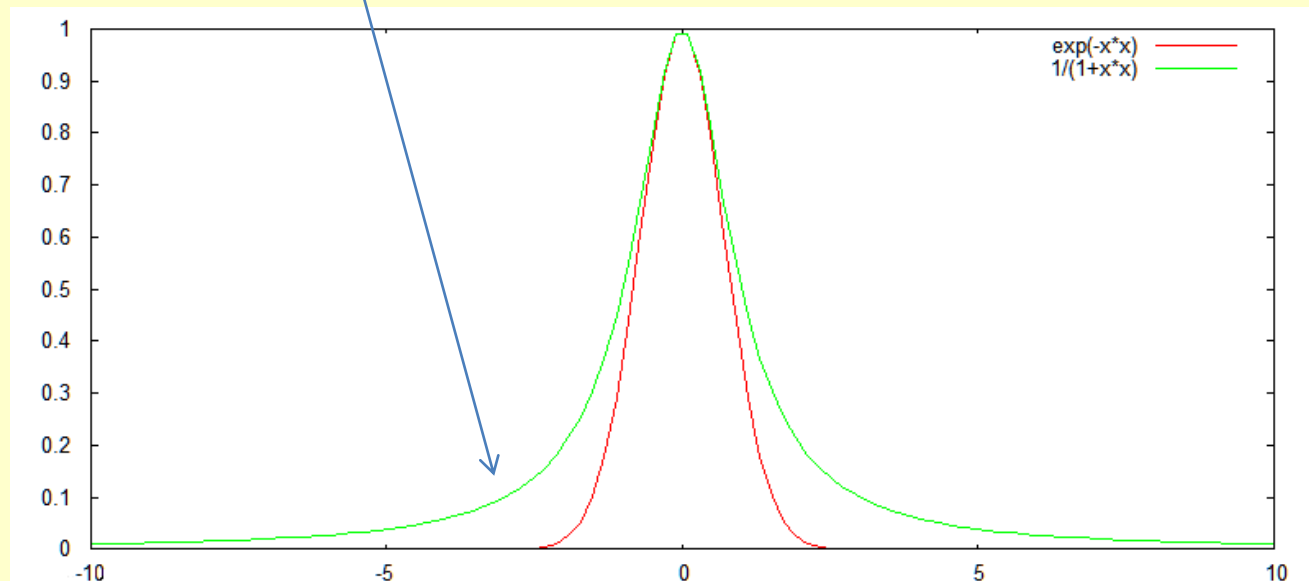
Optically thin

(weak) maser effects can also occur with CO under certain conditions of Temperature, density, cloud column density

Line emission (VI)

The **width** of a line is determined by

- Lifetime of transition: $1/A$
 - HI 21 cm line: $\Delta f/f = 2.85 \cdot 10^{-15} / 1.4 \cdot 10^9 \approx 10^{-24} !!!$
 - Line profile is Lorentzian shape ($\varphi \propto 1/(1+(\Delta f)^2)$)



Line emission (VI)

- Thermal and small-scale turbulent motions (Doppler effect) of the emitting gas

- **Gaussian** profile: width $\Delta f_D = f/c * \sqrt{\frac{kT}{\mu} + \xi^2}$

- hydrogen atoms at $T=100$ K give $v_D = 1$ km/s

- HI 21 cm line: 1 km/s $\rightarrow \Delta f_D = 4.7$ kHz

- Motions of HI clouds in the ISM: 10 km/s

- \rightarrow shape of HI 21 cm features reveals the kinematics and dynamics of the HI gas

Interpretation of Lines (I)

With Boltzmann formula define from level population ratio an **Excitation Temperature** (for this transition):

$$\frac{n_2}{n_1} = \frac{g_2}{g_1} \exp\left(-\frac{hf}{kT_{ex}}\right)$$

The ratio of emissivity and absorption is equal to the Planck function of this temperature:

$$S = \frac{j}{\kappa} = \frac{n_2 A_{21}}{n_1 B_{12} - n_2 B_{21}} = \frac{2hf^3}{c^2} \frac{1}{\frac{n_1 g_2}{n_2 g_1} - 1}$$

$$S = B(f, T_{ex})$$

Interpretation of Lines (II)

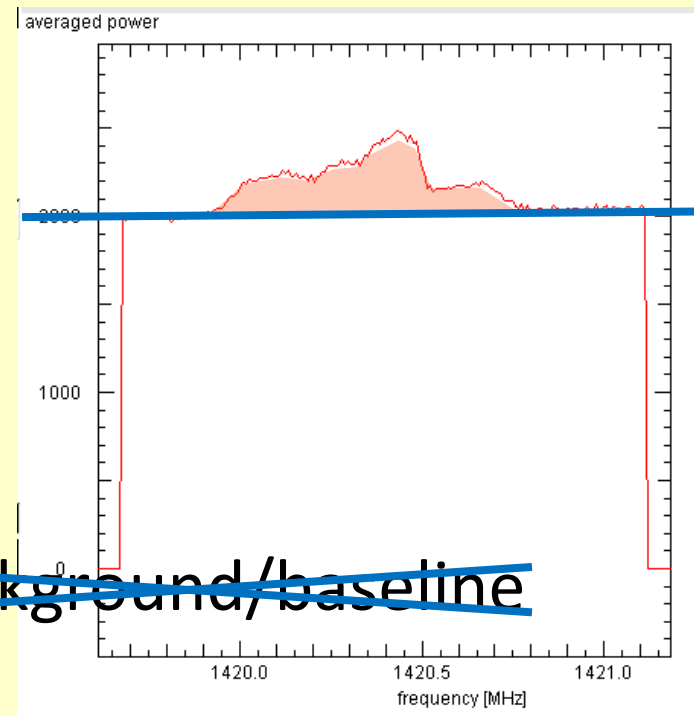
Optically thin line radiation lies 'on top' of any continuum or background:

- In terms of velocity shift:

$$I(\nu) = \tau(\nu) * B(T_{ex}) + \text{background/baseline}$$

- Integrate this emission excess over the entire line profile gives the total line emission

$$\int I(\nu) d\nu = B(T_{ex}) \int \tau(\nu) d\nu$$



Interpretation of Lines (III)

- Integrate optical depth over line profile

$$\begin{aligned}
 \int \tau(\nu) d\nu &= \iint \kappa(\nu, s) ds d\nu \\
 &= \frac{hf}{4\pi} \int (n_1 B_{12} - n_2 B_{21}) ds \int \cancel{\varphi(\nu) d\nu} = 1 \\
 &= \frac{hf}{4\pi} B_{12} (1 - e^{-hf/kT_{ex}}) \int n_1 ds
 \end{aligned}$$

- **Column density** of matter along the line of sight:
(N.B. $n = n_1 + n_2$)

$$N = \int n ds$$

Interpretation of Lines (III)

All together (with observed brightness temperatures):

$$\int T_B(\nu) d\nu = cte * N * (1 - e^{-hf/kT_{ex}}) T_{ex}$$

Thus, from the 21cm line one gets HI column density:

$$N_H = 1.822 \cdot 10^{18} \frac{1}{T_{ex}} \int T_B d\nu$$

in practical units: [atoms/cm²] column density, [K] temperatures, [km/s] velocity, for 21 cm line T_{ex} is called 'spin temperature'

Interpretation of Lines (IV)

Caveats:

- To get column density, one needs the **brightness** temperature
- This is equal to the measured antenna temperature only if the source is **well resolved!**